

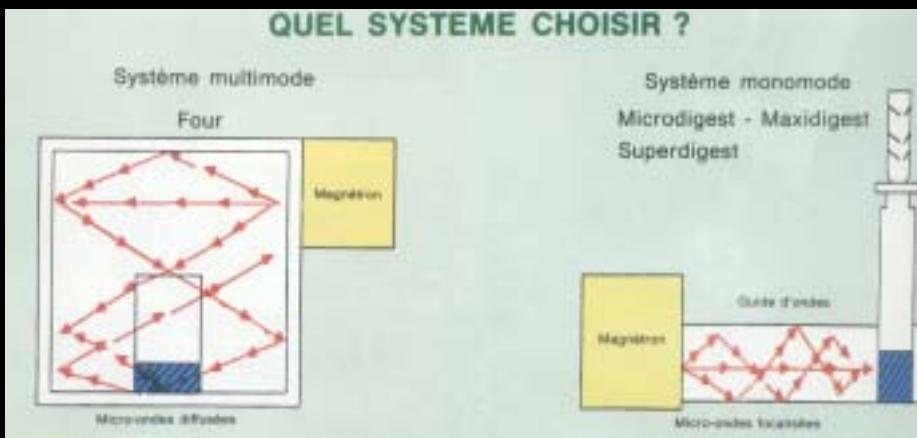
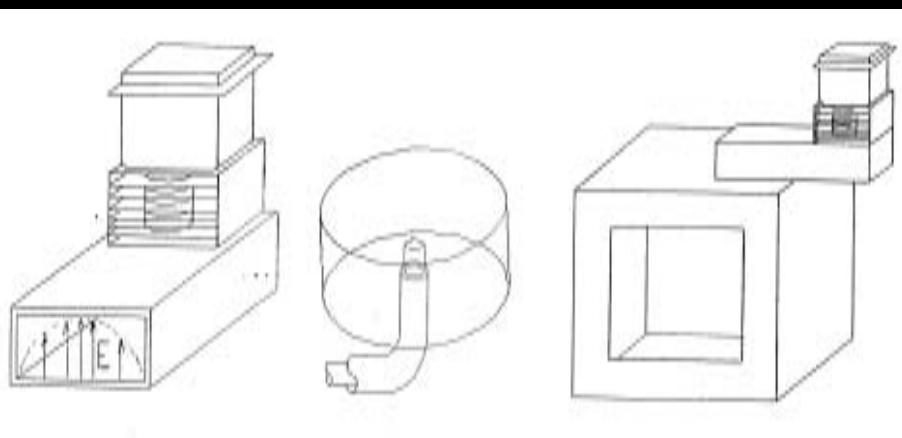
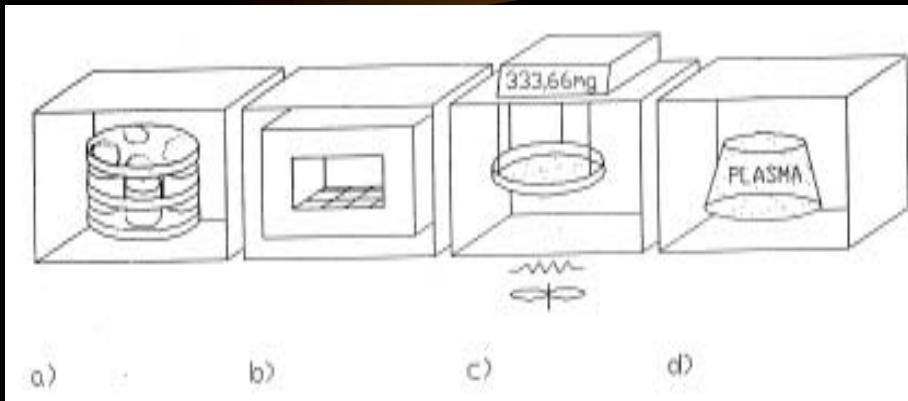
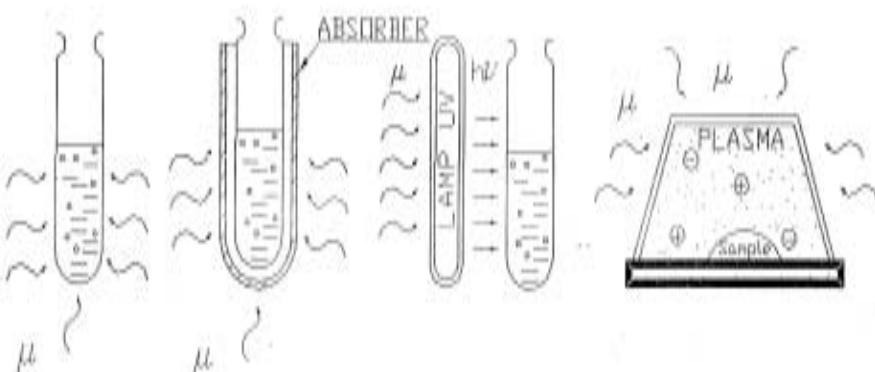
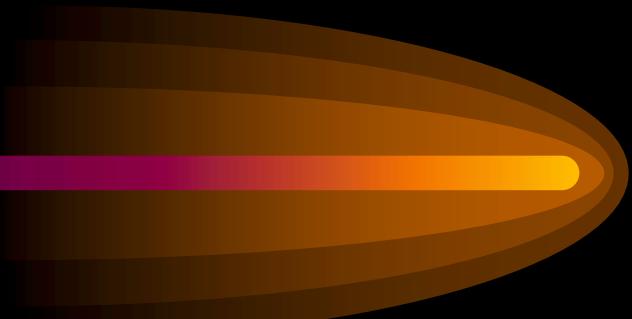
Selmar-GBC Slesin 2004

# Mikrofale w akredytowanym laboratorium analitycznym

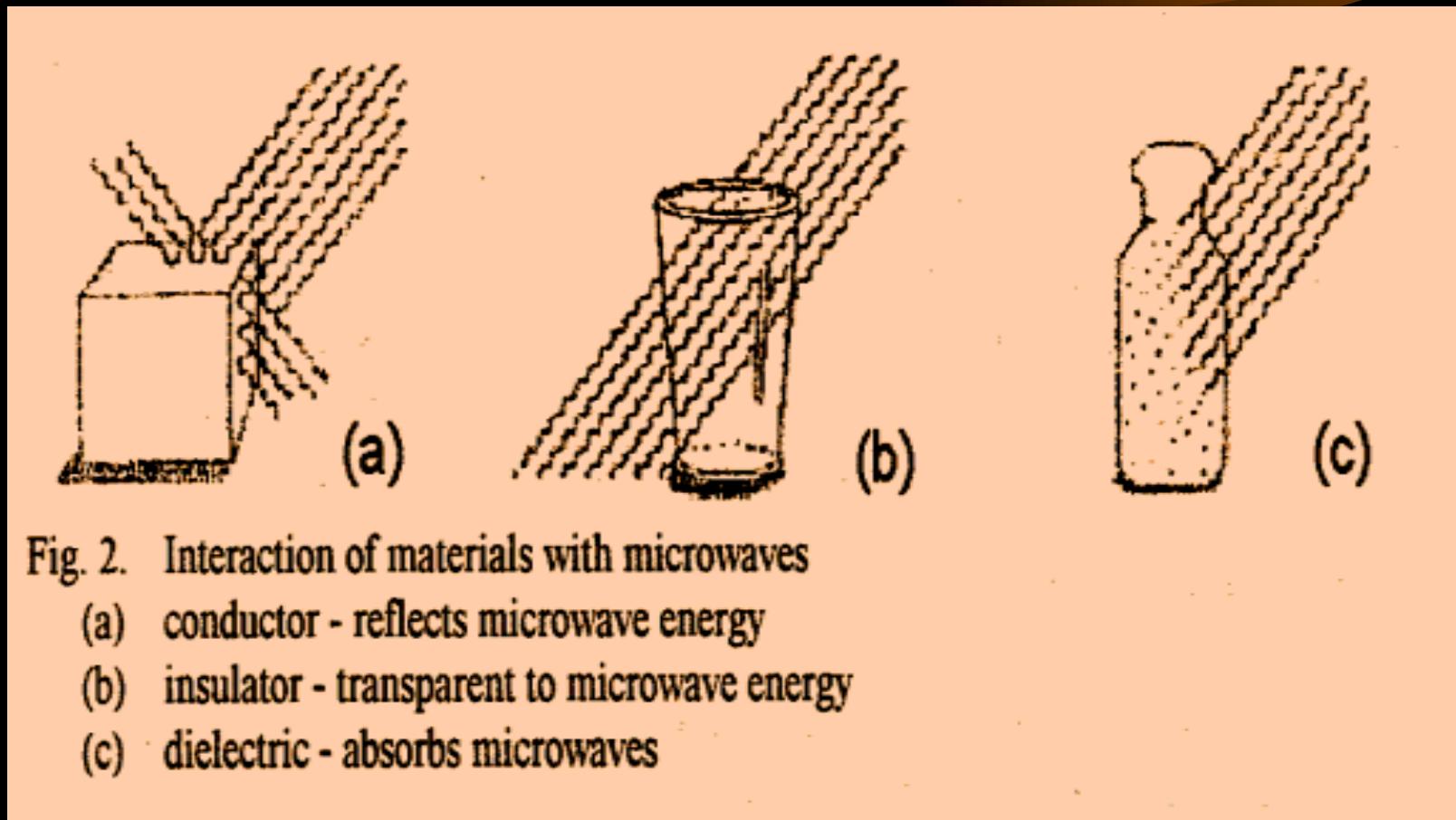
Edward Reszke  
ERTEC-Poland

May 2004

# Mikrofale w laboratorium



## Just for the beginning



# Piec mikrofalowy Milestone'a



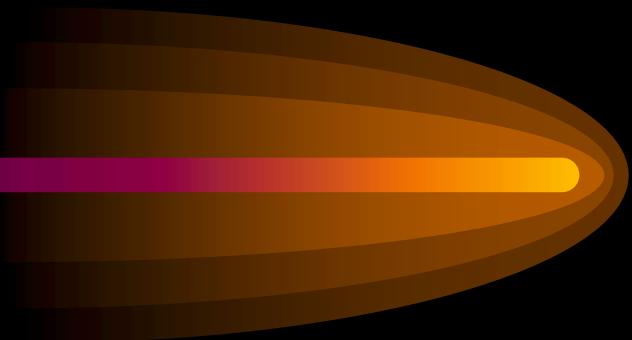
# Dry ashing of analytical samples

## Why you need a Milestone Pyro Ashing System

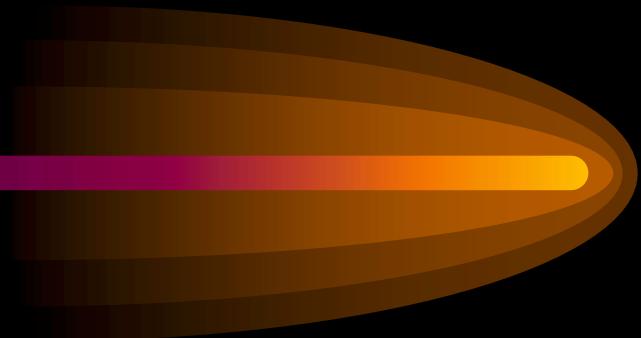
REDUCED ASHING TIMES: FROM HOURS TO MINUTES

Sample	Microwave Ashing Temperature (°C)	Microwave Ashing Time (minutes)	Traditional Ashing Time (minutes)
Pet food (ground)	575	21	90
Cat food	575	15	300
Polypropylene	650	22	80
PVC	900	15	120
Polyester	600	15	480
Polyurethane	900	15	120
Coal standard AR2782	750	20	120
Dried eggs yolk	925	25	240
Activated Coal	750	25	180
Flour	900	50	160
Salami	600	60	300

# Spopielarka plazmowa firmy Orbis Technologies Ltd



Karma dla zwierząt	ok. 10 godz.
Próbki krwi	2-4 godz.
Trzcinica cukrowa	ok. 4 godz.
Kakao	10-12 godz.
Filtr	1-2 godz.
Siano, trawa	10-12 godz.
Liście	4-8 godz.
Wątroba	ok. 12 godz.
Nasiona dyni	8 godz.
Olej sałatkowy	ok. 20 godz.
Mąka przemysłowa	2-3 godz.
Papierosy	ok. 1 godz.
Detergent – proszek	ok. 15 godz.
Grafit	3-4 godz.
Tkanina nylonowa	3-5 godz.
Papier	1-2 godz.
Włókna poliamidowe	2-3 godz.
Granulat poliamidowy	6-8 godz.



## Orbis-parametry pracy

Poliester	ok. 3 godz.
Polietylen	ok. 25 godz.
PVC	ok. 20 godz.
Próbki gleby	5-15 godz.

Komitet Techniczny ISO/TC-146 : *Jakość powietrza: Powietrze atmosferyczne. Oznaczanie włókien azbestowych. Metoda transmisyjnej mikroskopii elektronowej z pośrednim przeniesieniem próbki (2003).*

# Urządzenia do spopielania próbek azbestu w mikrofalowej plazmie tlenowej

**Edward Reszke<sup>1)</sup>, Marta Rożkowicz<sup>2)</sup>**

1) *Ertec-Poland 54-440 Wrocław ul. Rogowska 146/5 [www.ertec.pl](http://www.ertec.pl), [ertec@wp.pl](mailto:ertec@wp.pl)*

2) *Główny Instytut Górnictwa, 40-166 Katowice Plac Gwarków 1  
[soxmr@gig.katowice.pl](mailto:soxmr@gig.katowice.pl)*

Celem komunikatu jest przedstawienie projektu budowy pierwszego polskiego spopielacza plazmowego. Projekt dotyczy urządzenia do plazmowego spopielania całych filtrów z estru celulozy o wielkości porów 0,8 µm i średnicy Ø 25 mm wraz z próbką zawierającą włókna azbestu. W celu wykonania badań prototypu nawiązano współpracę badawczą pomiędzy firmą Ertec a Głównym Instytutem Górnictwa, który podjął się przebadania i optymalizacji charakterystyk nowego przyrządu przy pracy z próbками laboratoryjnymi materiałów azbestowych o różnych matrycach, czyli badań niezbędnych do atestacji nowego aparatu jako legalnego przyrządu laboratoryjnego spełniającego normy europejskie.

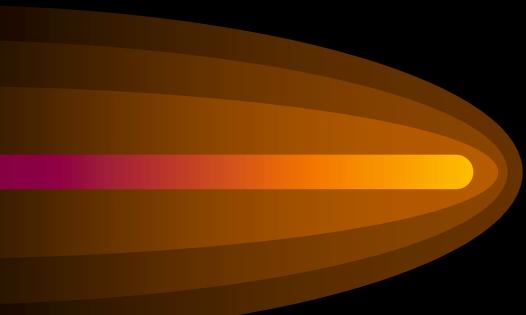
Przygotowywanie prób mikroskopowych materiałów zawierających azbest jest problemem, który pojawił się dopiero w ostatnich latach. Nowa analiza wymaga odkrywania struktury niebezpiecznych włókien azbestu i ich rozróżniania w masie

# Prototyp plazmowej spopielarki próbek azbestowych



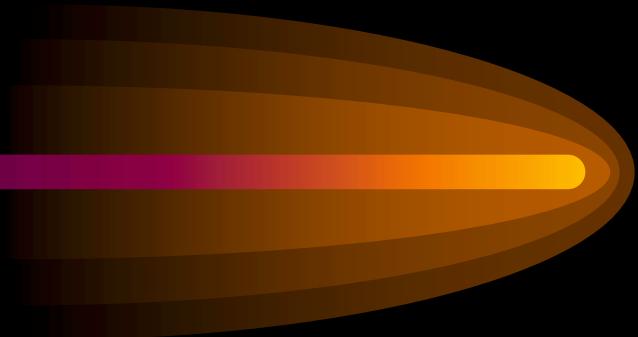
# Wnęka prototypowej spalarki





**Widok podajnika  
próbki  
(filtr celulozowy z  
ACM)**

# Najnowsze źródło wzbudzenia MIP z wnęką typu TEM



An example of MW system with integrated power source, coaxial feeder and two stub tuner placed between MW generator and applicator (a microwave cavity with TEM mode for MIP excitation)

# Otwarte systemy mineralizacji „na mokro”



Maxidigest MX350  
Prolabo



Star 2  
CEM

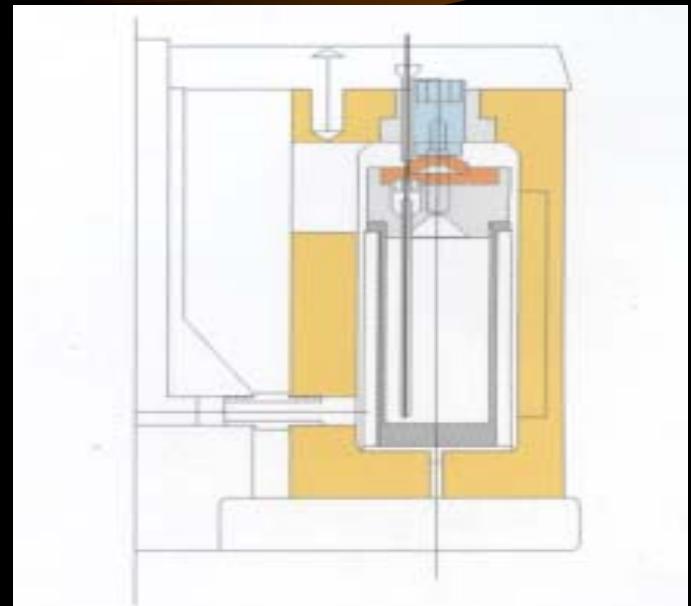
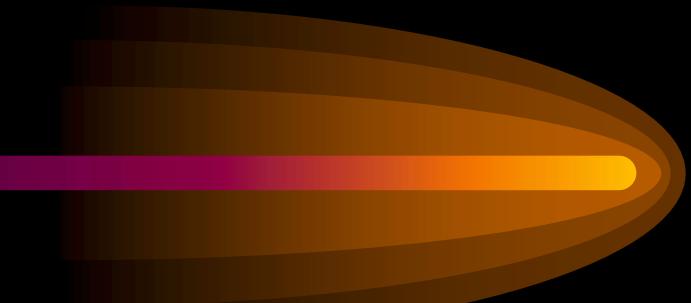


UniClever BM-1o  
Plazmatronika

# High pressure MW systems



Ethos plus  
from Milestone



*State of the art, rugged, “quick disconnect” thermocouple temperature control*

digestion vessel  
Ethos

# High pressure system



Plazmatronika-  
Nano 2000

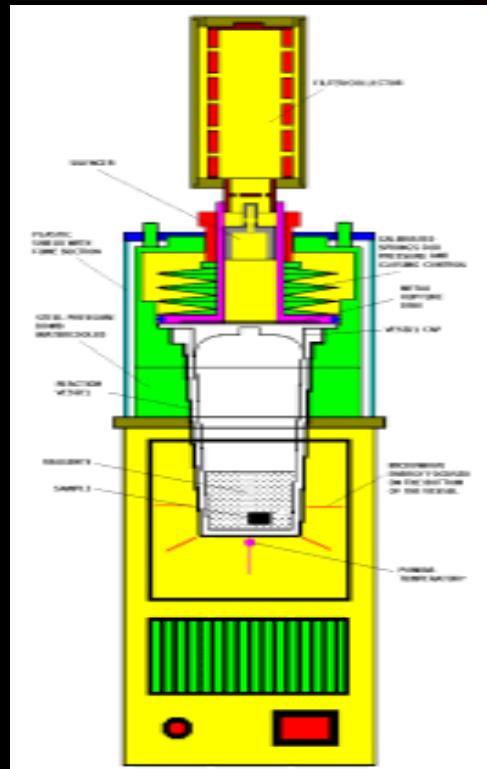


Reactor/mineralizer Ertec

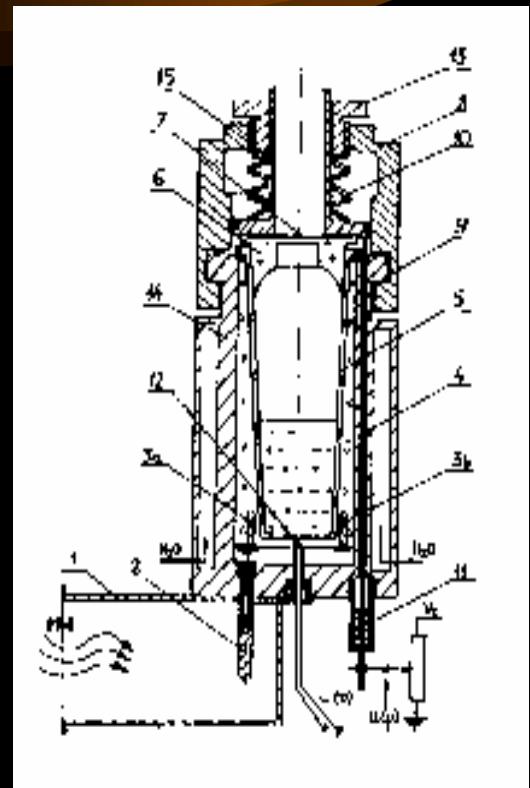
# Concept of a novel closed system with self- matching feature



Ertec system Magnum

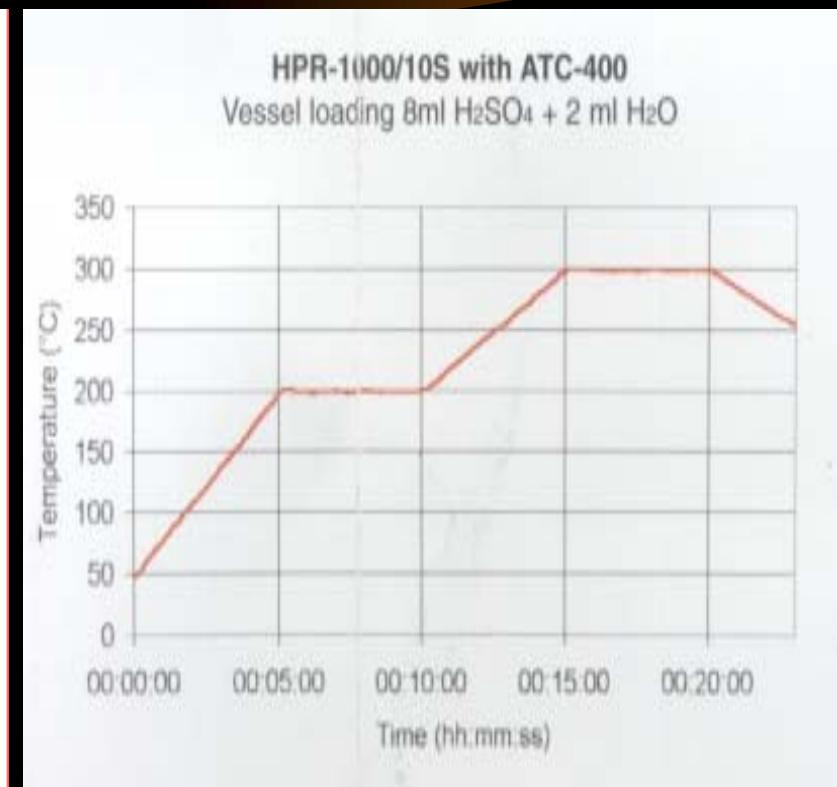
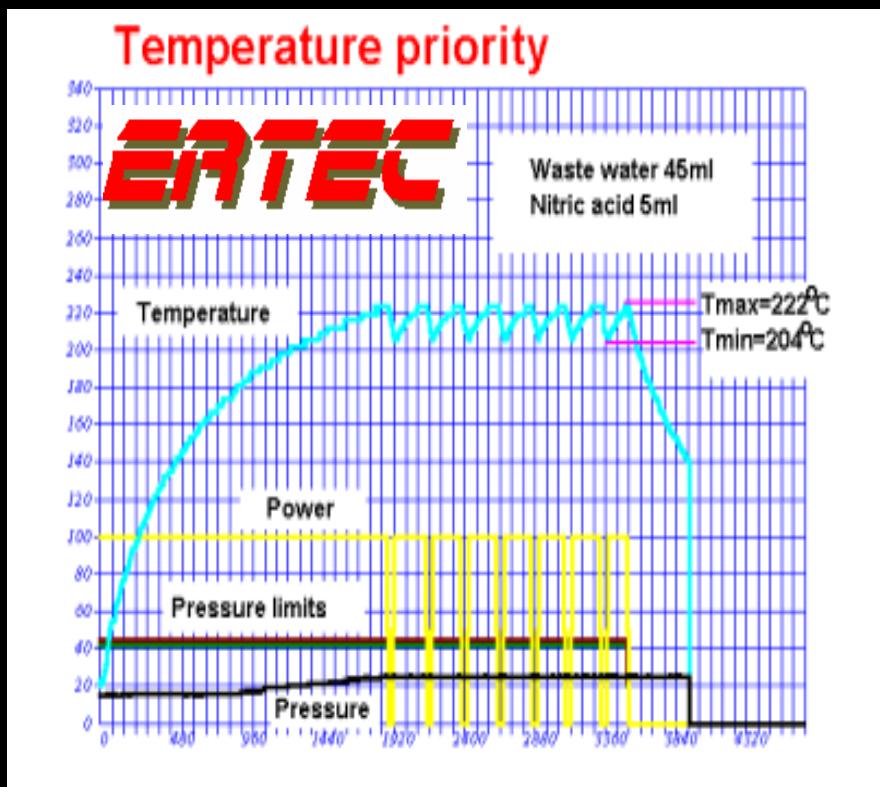


Construction of ERTEC  
mineralizer/reactor

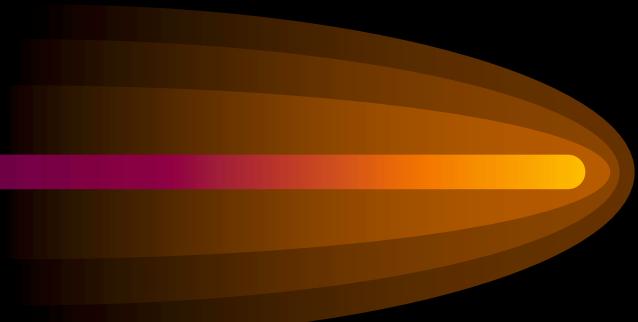


Magnum cross section

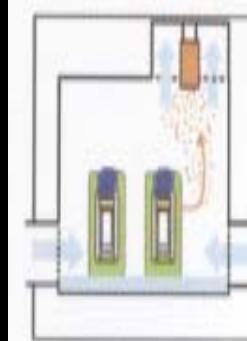
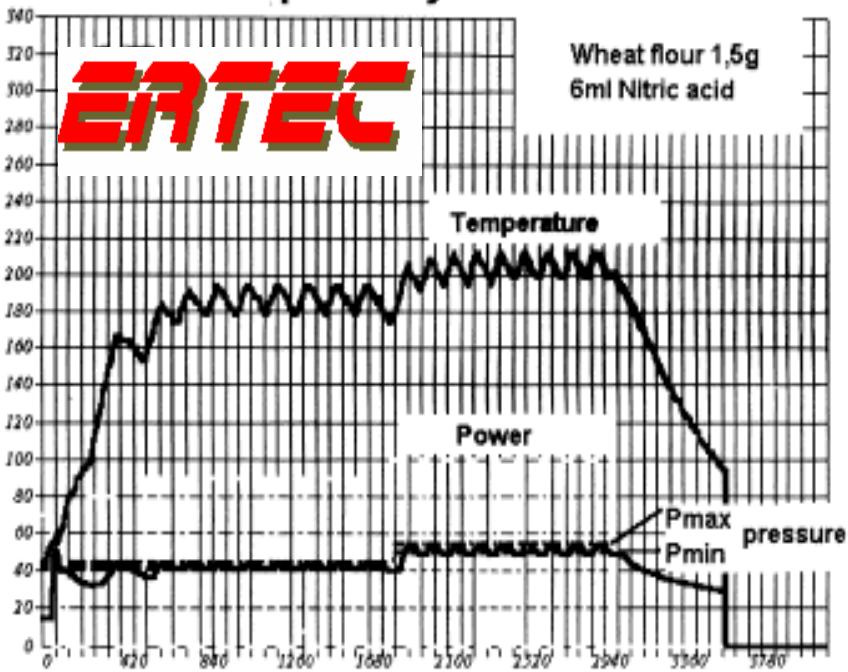
# Temperature measurements



# Measurement and control of pressure



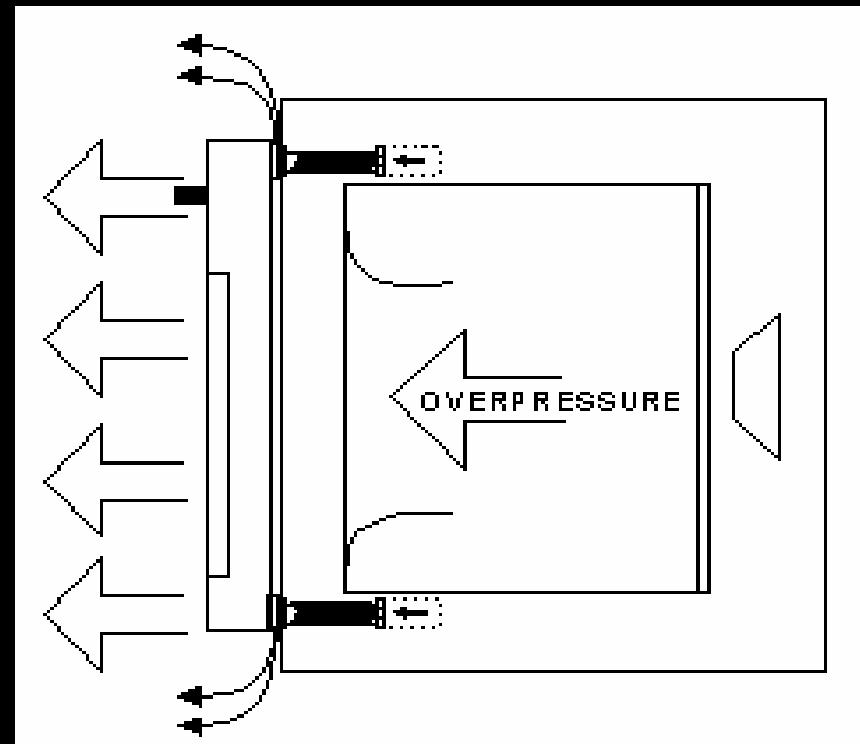
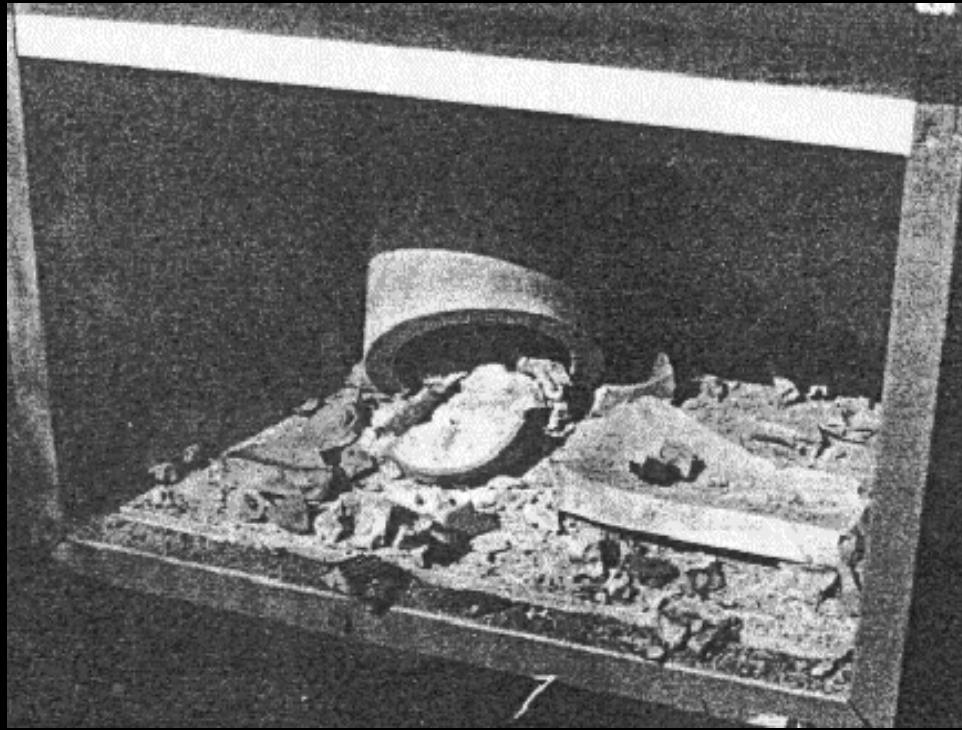
## Pressure priority



*QP for pressure limit control*  
Another first from Milestone's R&D Department. This unique sensor continuously monitors the level of acid vapors in the microwave cavity.

Once over a preset level, the sensor automatically responds by adjusting the microwave power. All vessels in the cavity are monitored simultaneously and independently to effectively limit the pressure in each vessel, thereby preventing overpressure and venting.

# Bezpieczne konstrukcje



# Rozpuszczanie materiałów nieorganicznych

**Table 7. Dissolution speed of „conventional” and microwave closed approaches**

	Conventional Teflon bomb		Microwave digestor
Matrix	(h)	(min)	(min)
$\text{Al}_2\text{O}_3$	24	1440	20
AlN	2	120	15
BN	5	300	15
$\text{Si}_3\text{N}_4$	6	360	15

# Digestion efficiency

Table 5 Effective of destruction of the organic matrix in the study systems decomposition

Cert. Ref. Material	Microwave System	CRM-total C [ % ]	Degree of C removed C [ % ]	RSD of RCC [%]
Pig Kidney 186	Paar Physica	48,96	99,25	35,9
	MLS MEGA		97,54	39,7
	Prolabo T		98,93	21,4
	Prolabo S		94,25	40,8
Dogfish DORM 2	Paar Physica	44,20	98,87	34,2
	MLS MEGA		97,07	27,6
	Prolabo T		98,70	35,0
	Prolabo S		94,00	38,4
Mussel	Paar Physica	43,88	99,01	32,4
Tissue 278	MLS MEGA		97,70	42,6
	Prolabo T		99,21	10,0
	Prolabo S		95,26	39,3
Brown Bread	Paar Physica	45,60	99,08	28,9
BCR 191	MLS MEGA		97,14	31,8
	Prolabo T		98,94	11,4
	Prolabo S		95,21	35,5
Aquatic Plant	Paar Physica	37,82	98,92	30,9
BCR 60	MLS MEGA		97,65	34,1
	Prolabo T		98,87	22,6
	Prolabo S		95,11	38,7

RCC - residual carbon content

# Digestion reagents

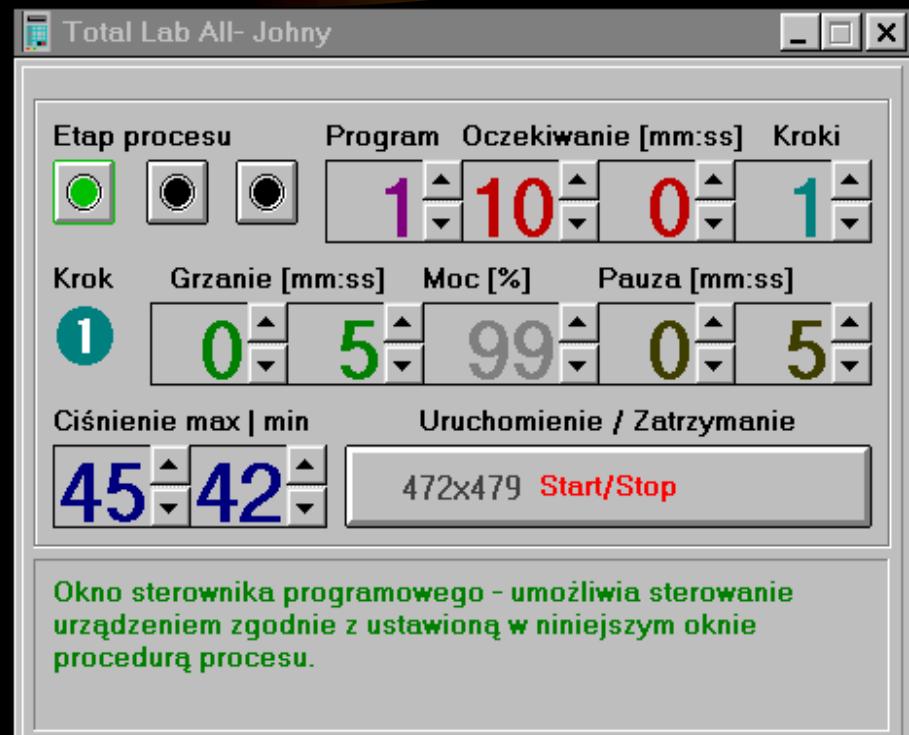
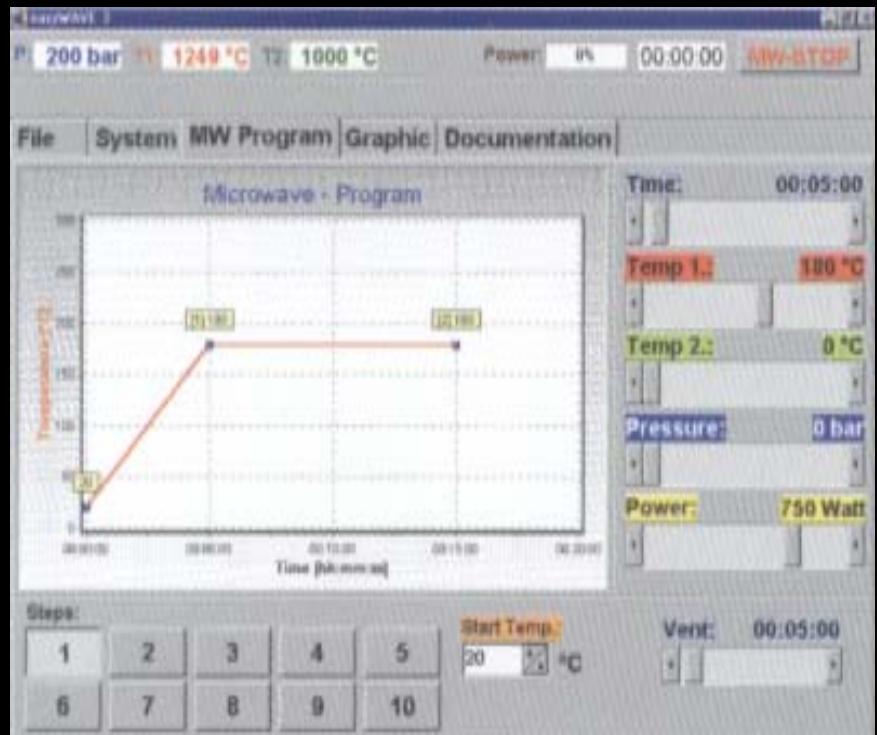
Solution of reagents	Matrix	Uwagi
HNO <sub>3</sub>	Biological materials	incompletye digestion at atmospheric pressure
HNO <sub>3</sub> + H <sub>2</sub> O <sub>2</sub>	Biological materials	Effective digestion of small samples
HNO <sub>3</sub> + H <sub>2</sub> SO <sub>4</sub>	general use	Frequently used but promotes losses of easily volatile elements such as As, Ge, Hg, Se
HNO <sub>3</sub> + HCl	general use	aqua regia
HNO <sub>3</sub> + HClO <sub>3</sub>	Biological materials	High oxidation potential, low loss of As, Hg, Se, In and so on .., safe
HNO <sub>3</sub> + HClO <sub>4</sub>	Biological materials	Effective decomposition , low loss of Pb
HNO <sub>3</sub> + HCl + HF	general use	-
HNO <sub>3</sub> + HF	general use	-
HNO <sub>3</sub> + HClO <sub>4</sub> + H <sub>2</sub> SO <sub>4</sub>	general use	Temperature regime must be controlled , loss of As, Hg, Fe Sb
HF	innorganic materials	-
H <sub>2</sub> SO <sub>4</sub> + HClO <sub>4</sub>	general use	Small samples , danger of explosion
H <sub>2</sub> SO <sub>4</sub> + H <sub>2</sub> O <sub>2</sub>	general use	Losses of As, Hg, Ge, Ru, Se and many other volatile elements
HClO <sub>4</sub>	Biological materials	Strong oxidizer, danger of explosion
H <sub>2</sub> O <sub>2</sub> + Fe <sup>3+</sup>	Biological materials except oils, fats and grees	Decomposition with OH* radicals, low temperature of decomposition, good for large samples, no loss of elements

# Material of digestion vessels, memory effects



Material	Melting point	Maximal . T[°C]	tg x10 <sup>4</sup>
Teflon PFA	302	260	2,1
Teflon FEP	252-262	204	2,1
Halon PTFE	320	260	2,1
Polymethylpentane	240	175	2,1
Polyethylene	120-135	71-93	2,2-2,3
High press.polyethylene	146	121	2,25
Polypropylene	168-171	100-105	2,24-2,4
Polystyrene	242	82-91	2,7-3,1
Polycarbonate	241	121	2,9
NYLON 6	216	102	3,4
Quartz glass	1665		3,8-4,1
Borosilicat.glass	1080		6,3-6,8
Phenol resin	rozkladaju sa	120-190	4,1-5,0

# Software and data aquisition



## Digestion of difficult organic samples with dry mass over 1g

**ERTEC**

Sample	Mass [g]	Reagents [ml]	Digestion procedure power [%] pressure set. [at]	Pressure reached (oscillation) [at]	Total digestion time [min]
Green coffee	1,5	6 HNO <sub>3</sub>	5'/60% 20-17at; 5'/100% 45-42at 10'/0%	45 (45-42)	20
Tea fix	1,5	6 HNO <sub>3</sub> , 2 H <sub>2</sub> O	5'/60% 20-17at; 5'/80% 45-42at; 10'/100%	45 (45-42)	20
Rice	2,5	8 HNO <sub>3</sub> , 4 H <sub>2</sub> O	10'/60% 20-17at; 10'/0% 45-42at; 10'/0%	105 (45-42)	30
Goulash soup	1,5	8 HNO <sub>3</sub> , 4 H <sub>2</sub> O	10'/60% 20-17at; 10'/100% 45-42at; 10'/0%	74 (45-42)	30
Dry mushrooms	2	8 HNO <sub>3</sub> , 4 H <sub>2</sub> O	10'/60% 20-17at; 10'/100% 45-42at; 10'/0%	45 (45-42)	30
Peanut	1,5	8 HNO <sub>3</sub> , 4 H <sub>2</sub> O	10'/60% 20-17at; 10'/100% 45-42at; 10'/0%	45 (45-42)	30
Poppy seed	2	8 HNO <sub>3</sub> , 4 H <sub>2</sub> O	10'/60%; 10'/100%; 10'/0%	82 (45-42)	30

# Efficiency of digestion systems

Dzienna zdolność przerobowa różnych systemów do mineralizacji			
Rodzaj próbek	Masa próbki [sucha]	Wysokociśnieniowy system modułowy [2 moduły]	Systemy mikrofalowe z wnęką wielomodową [6 stanowisk]
Organiczne	do 0,5 g	64 próbek dziennie	48 próbek dziennie
	do 1 g	48 próbek dziennie	zwykle niemożliwe
	do 2 g	24 próbek dziennie	niemożliwe
	do 5 g	12 próbek dziennie	niemożliwe
Nieorganiczne	do 0,5 g	32 próbek dziennie	30 próbek dziennie
	do 1 g	24 próbek dziennie	18 próbek dziennie
	do 2 g	12 próbek dziennie	czasem niemożliwe
	do 5 g	6 próbek dziennie	niemożliwe
Mieszane	do 0,5 g	48 próbek dziennie	36 próbek dziennie
	do 1 g	36 próbek dziennie	24 próbek dziennie
	do 2 g	18 próbek dziennie	zwykle niemożliwe
	do 5 g	10 próbek dziennie	niemożliwe

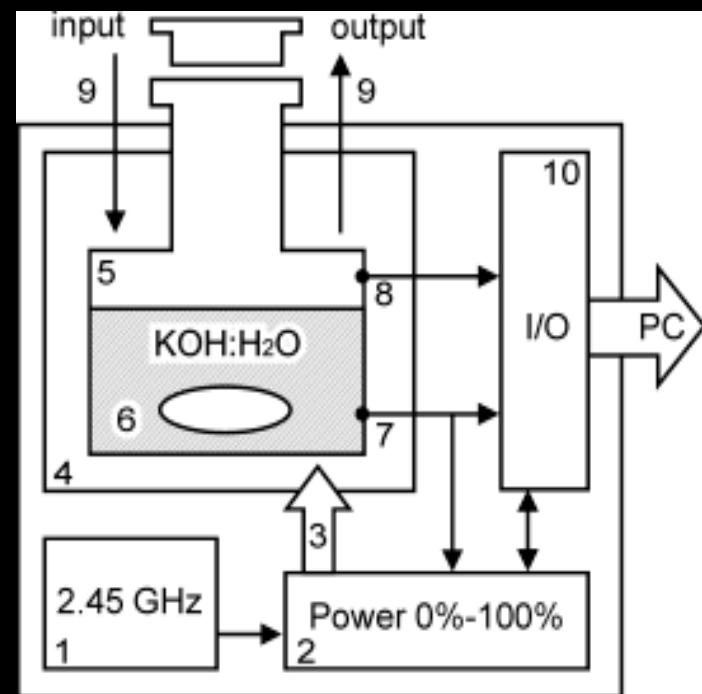
# MW vs Conventionally driven reactions

Kind of reaction:	MW reactor		Conventional	
	yield	time	yield	time
Diels-Alder Condensation (anthracen with dimethyl ester of fumaric acid)	87%	10min	67%	4h
Cyclocondensation reaction - synthesis of pyrimidobenzimidazolines - synthesis of derivatives of cumarine	38-86%  75-90%	15-30min  1-10min	brak  60-90%	48h  6h
Reformation of Claisen aromatical alkilic ethers	100%	90s	100%	12min
Alkilation of salts of carboxylic acids - production of octyl acetate - production of hexadecyl acetate	99%  96%	2min  2min	2%  brak	2min  20h
Reakcje o-alkilation with alkilic dihalogens	80-90%	5min	90%	30min
Cation Reformation , pinakolin reformation after adsorption of pinakol onto montmorylonit	99%	15min	99%	15h
Oxidation reactions				
- oxidation of I- and II-order alcohols leading to aldehydes and ketons in the presence of MnO <sub>2</sub>	50-80%	3-7min	25-40%	11h
- oxidation of I- and II-order alcohols to acids and ketons against H <sub>2</sub> O <sub>2</sub>	60-97%	10-20min	83-96%	4h
- epoxidation of double bonds against H <sub>2</sub> O <sub>2</sub>	91-98%	100min	54-65%	100min
One-step precipitation of magnetite	100%	10 min	100%	60 min
Precipitation of zirconia powders	100%	20 min	100%	240 min
Precipitation of zinc oxides	100%	3 min	90%	40 min

# Etching Microwave Silicon

Novel method, named EMSi (Etching Microwave Silicon) a Teflon® reaction vessel containing silicon substrates immersed in the 3 - 5 M KOH solution is positioned inside the microwave resonator and irradiated by 2.45 GHz microwaves. Etching rates for (100) planes:  $V_{100}$  equal to 5 ÷ 10  $\mu\text{m}/\text{min}$ , in almost cold ( $60^\circ\text{C} \div 70^\circ\text{C}$ ) solutions have been obtained. New, microwave enhanced etching of silicon enabled fast fabrication of the 3D silicon micromechanical structures.

EMSi etching: a) scheme of the stand: 1-microwave generator, 2-power supply unit, 3-connector, 4-single or multimode resonator, 5-reaction chamber, 6-silicon substrate, 7-temperature sensor, 8-pressure sensor, 9-cooling water, 10-I/O, b) process characteristics: pressure, microwave power and temperature versus time.

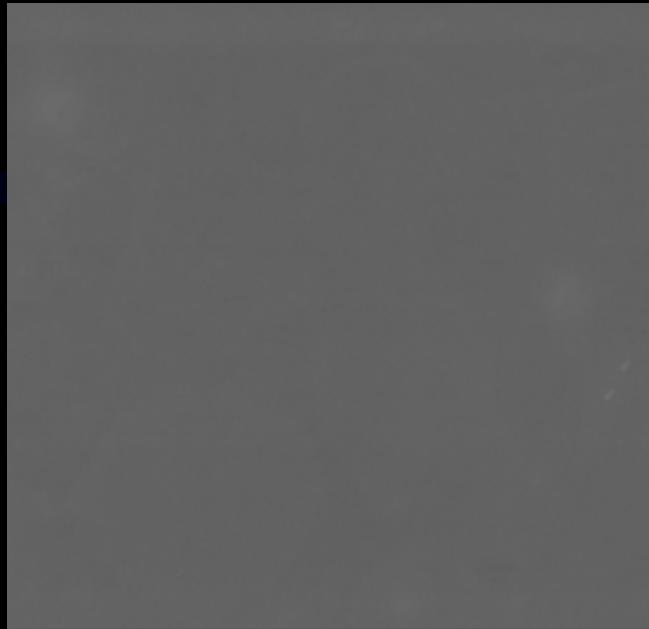


# All stainless steel reactor for etching experiments

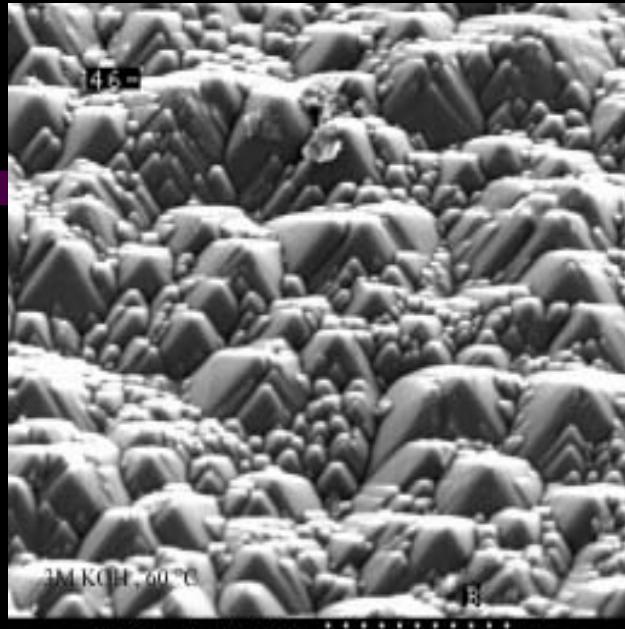




Standard procedures		Microwave-enhanced	
$\text{M}$	$\text{M}$	$\text{M}$	$\text{M}$
low etching rates $V_{(100)} \approx 1 \mu\text{m/min}$ at $80^\circ\text{C}$		high etching rates $V_{(100)} \leq 10 \mu\text{m/min}$ at $80^\circ\text{C}$	
etching rates depend on etchant's temperature		etching rates depend on microwave power "pumped" into the solutions	
hillocks formed in low concentration KOH or (and) low temperature		no hillocks	
good anisotropy (10 M KOH $\sim 40\div50$ ) higher for stronger solutions		sufficient anisotropy 30÷20 higher for weaker solutions	
batch process		single wafer process, batch process possible	
long (hours)		short ( $< 60$ minutes)	
etching rates at $60^\circ\text{C}$ negligible		etching rates at $60^\circ\text{C}$ high ( $\sim 5 \mu\text{m/min}$ )	
etching rates in 3 M KOH negligible		etching rates in 3 M KOH high ( $\sim 5 \mu\text{m/min}$ )	
$\text{SiO}_2$ , $\text{Si}_3\text{N}_4$ mask		$\text{Si}_3\text{N}_4$ mask only; selectivity 1:10000	
60°C process impossible		60°C process good	



161225 10KV X5.00K 6.0um



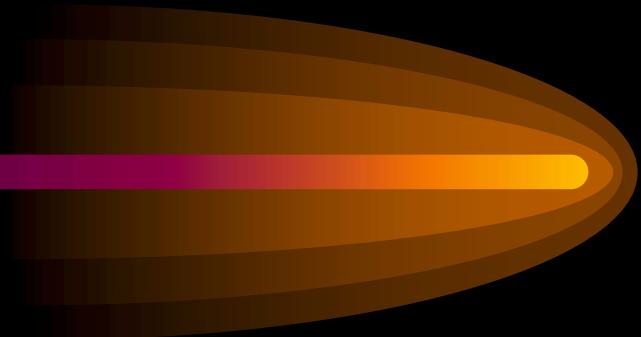
000138 7.0KV X3.00K 10.0um

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) new etching EMSi, open system,  $V_{(100)}=5,1 \mu\text{m}/\text{min}$ ,

b) conventional etching,  $V_{(100)}=0,4 \mu\text{m}/\text{min}$ , Photography SEM x5000, x3000

# Bezpieczeństwo użytkowania urządzeń mikrofalowych



1. Dyrektywa niskonapięciowa
2. Dyrektywa ciśnieniowa dot. Prostych zbiorników ciśnieniowych  $pV < 10.000 \text{ bar cm}^3$
3. Dyrektyw kompatybilności EM ( $\Pr < 5 \text{ mW/cm}^2$ )
4. Dokumentacja techniczna
5. Dokumentacja pomiarów i prób technicznych
6. Certyfikat CE producenta ( lub/i -jednostki akredytowanej)

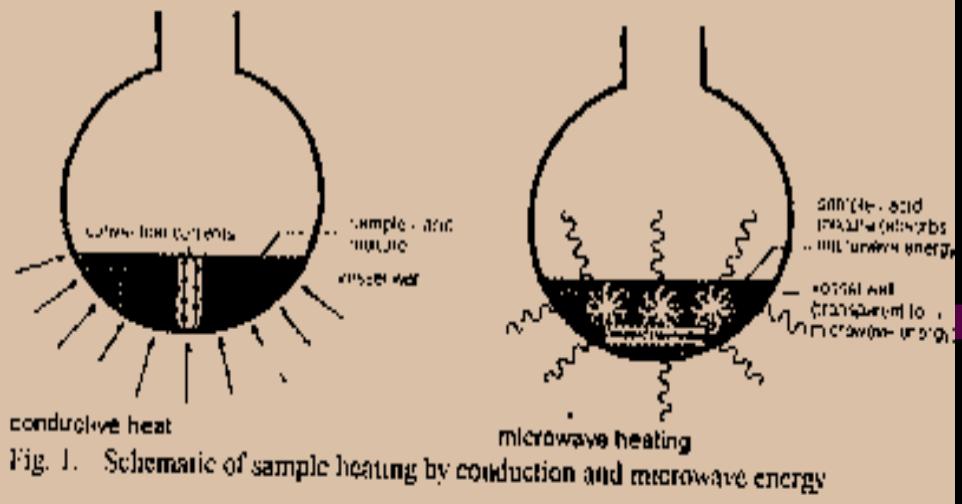
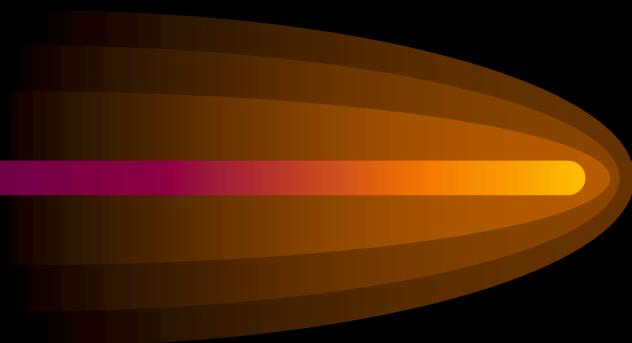


Fig. 1. Schematic of sample heating by conduction and microwave energy



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